

# Download Chemistry Valence Electrons Answer Key

Number of Valence Electrons. As mentioned before, sodium has one valence electron (that 3s electron), which is one reason why it is so reactive and unstable. Print Atoms Worksheets Click the buttons to print each worksheet and associated answer key. Sr > Ca > Mg > Be Atomic radii increase going down a group, because successively larger valence-shell orbitals are occupied by electrons. For example, strontium has electrons in the fifth shell, which contains much larger orbitals than the fourth, third, second, or first shells. Key Concepts . The electrons on the outermost energy level of the atom are called valence electrons. The valence electrons are involved in bonding one atom to another., Chemistry Valence Electrons Answer Key.

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